

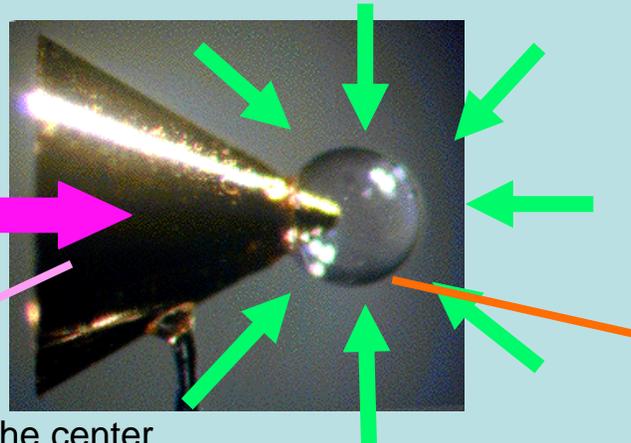


Low density aerogel capsule for FIREX (Fast Ignition Realization EXperiments) project –control of nanostructure–

PW(10^{15} W) for heating
1 beam / 300 J
1.053 μm / 0.6ps

Au cone

30 ° open angle (the picture: 60deg)
Thickness of the cone top: 5 μm
Distance of the cone top: 50 μm from the center



GXII for implosion
9 beams / 2.5 kJ/0.53 μm

1.2ns Flat Top w/ RPP

Deuterated polystyrene
shell 500 μm ϕ /6-7 μm t

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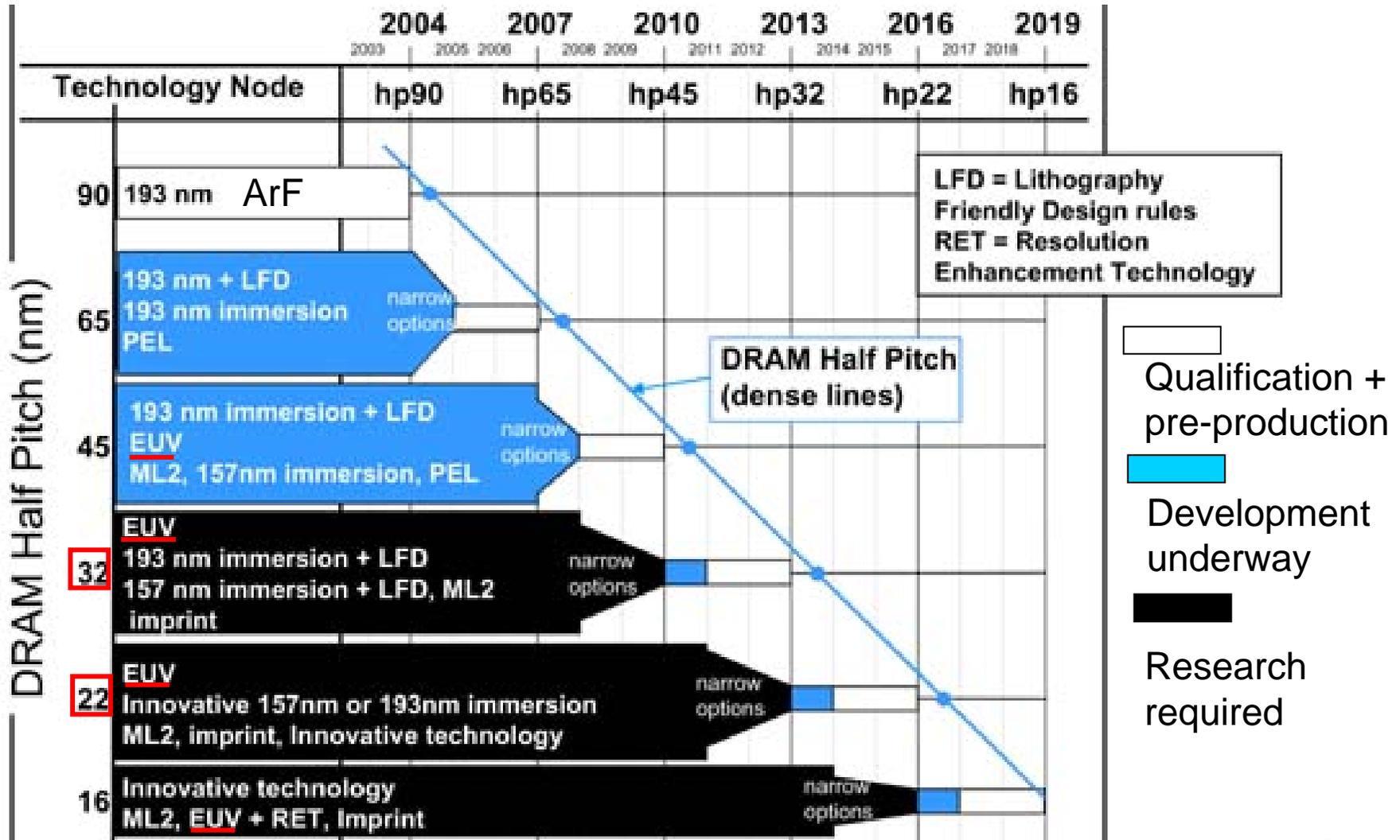


Outline

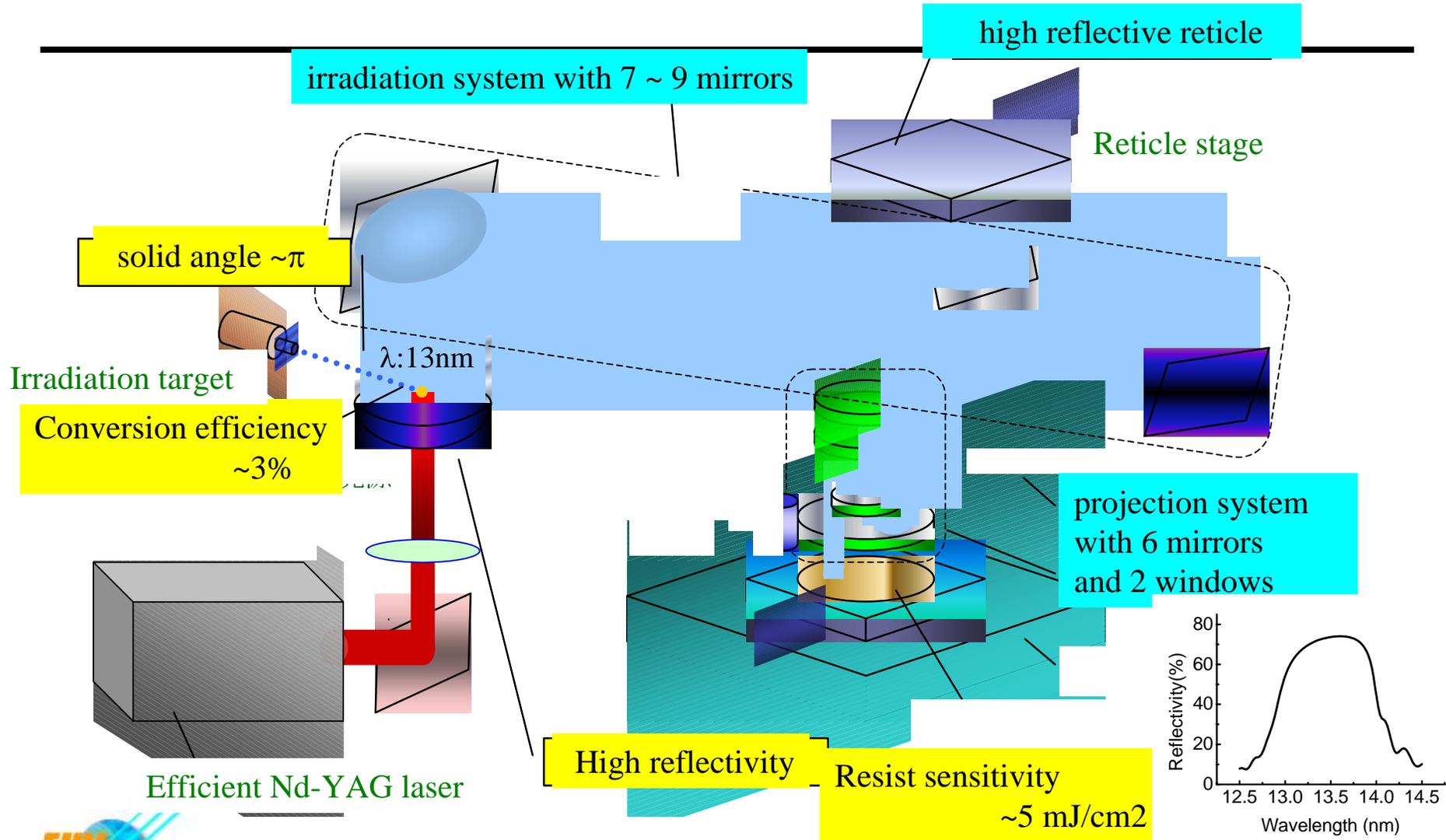
1. Low density Sn and Au using template, electrospinning
2. High rep target ----necessity of minimum mass target
3. Organic aerogel capsule
4. Discussion of nanostructure control



Lithography load map shown in ITRS 2004 update



Extreme UV (EUV) lithography system design

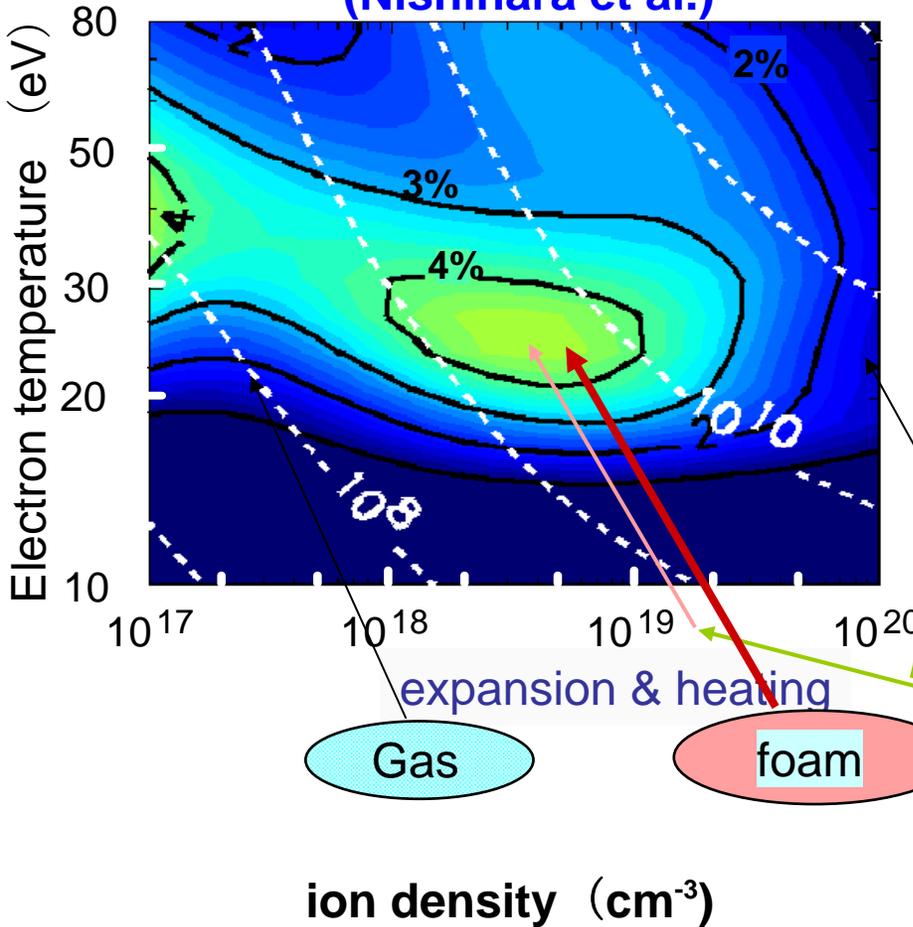


Theoretical analysis gave a mapping of conversion efficiency

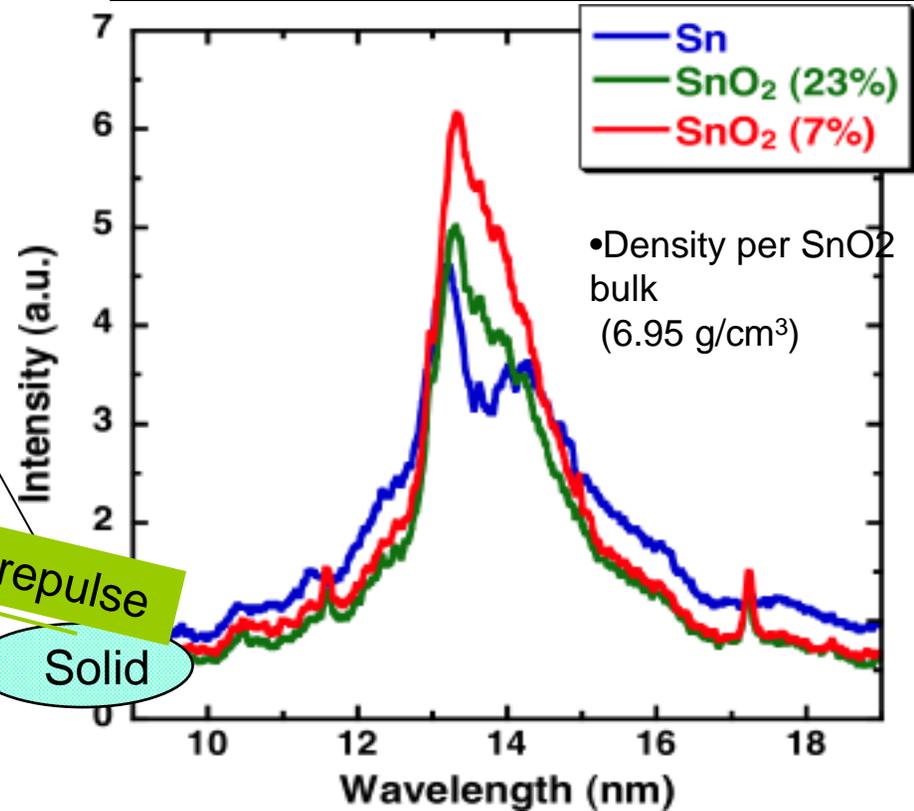
30 eV, $10^{18} \sim 10^{19} \text{ cm}^{-3}$



Conversion efficiency for EUV emission (Nishihara et al.)



Energy normalized EUV spectra

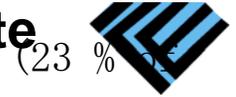


10 ns, 1064 nm

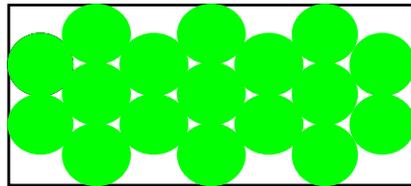
Appl. Phys. Lett., **88** (16), 161501 (2006).



Low density tin oxide ($d=1.5\text{g/cm}^3$) using nanotemplate

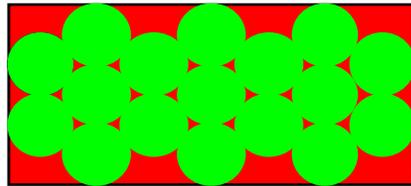


bulk crystal)



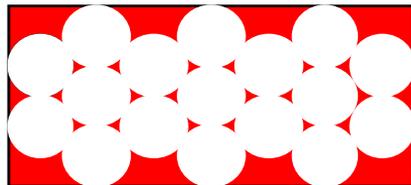
Polystyrene particles were aligned to be closed packing.

SnCl_4 ↓



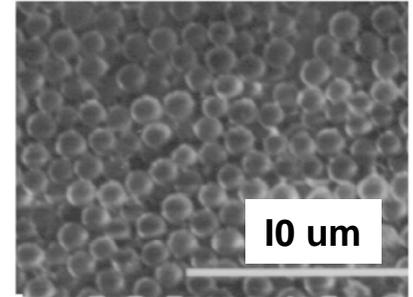
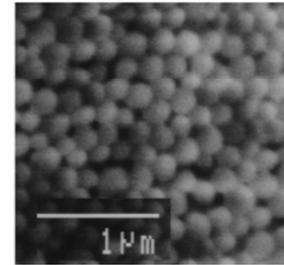
PS particles were immersed in liquid tin chloride. Tin chloride was hydrolyzed to be tin oxide.

400°C ↓

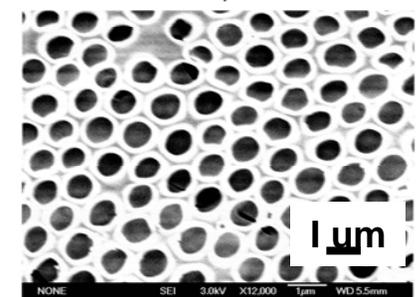
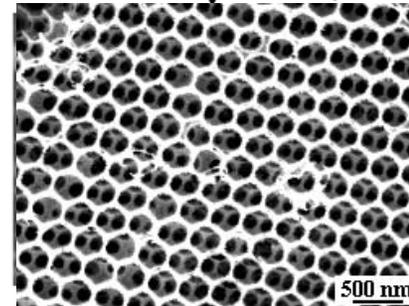


PS particles were decomposed by heating. Porous tin oxide film was obtained.

Polystyrene nanoparticles



nanoporous tinoxide



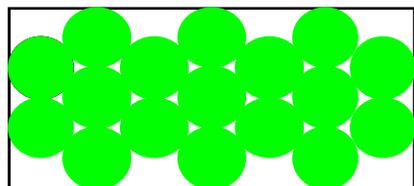
Pore size was well controlled by the template spheres.



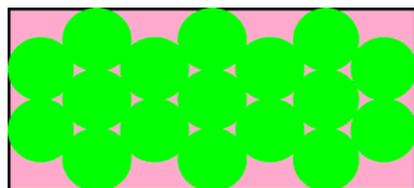
Low density tin oxide ($d=0.5\text{g/cm}^3$) using nanotemplate (7 % of bulk crystal)



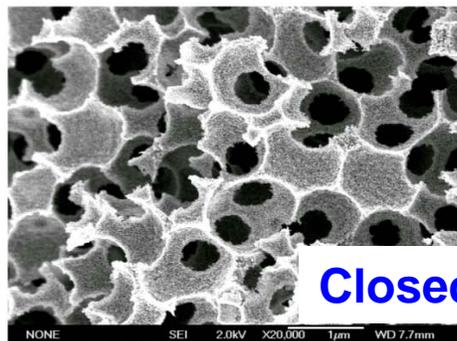
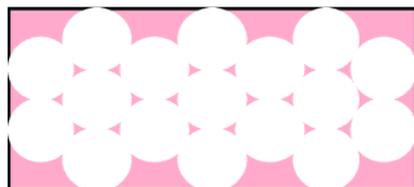
$$n = \text{EtOH}_{\text{mol}} / \text{SnCl}_4_{\text{mol}}$$



$\text{SnCl}_4/\text{EtOH}/\text{H}_2\text{O}$

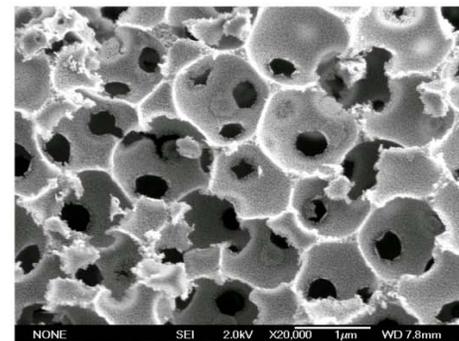


400°C

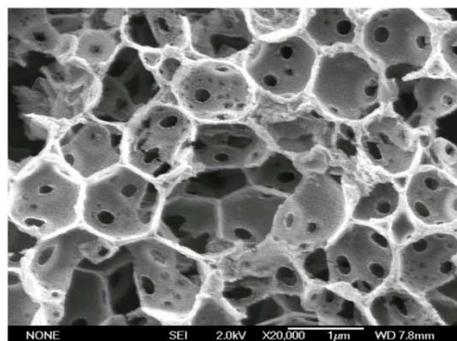


$n=2$

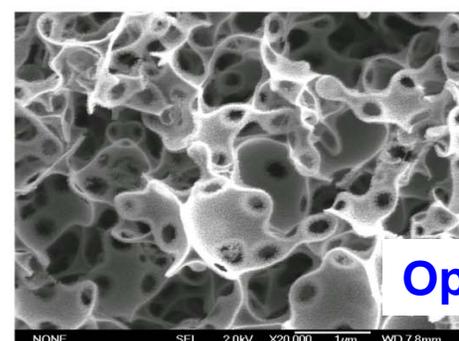
Closed cell



$n=4$



$n=6$



$n=10$

Open cell

There were various morphologies.

Q.C.Gu et al., Chem. Mater., **17** (5), 1115-1122, (2005).



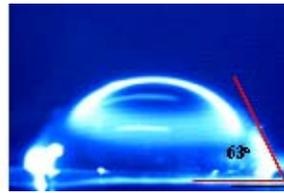
Volume template vs. Surface template



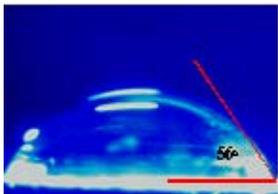
Contact angle on polystyrene film



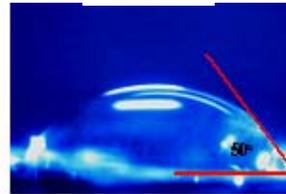
n=2



n=4



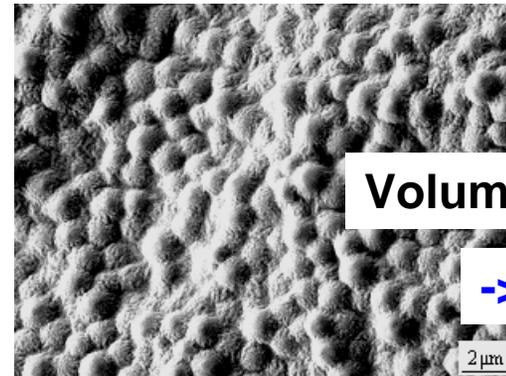
n=6



n=10

More EtOH (n), lower contact angle
(higher affinity for PS)

SEM images before heating at 400 °C



n=2

Volume template

-> Closed cell



n=10

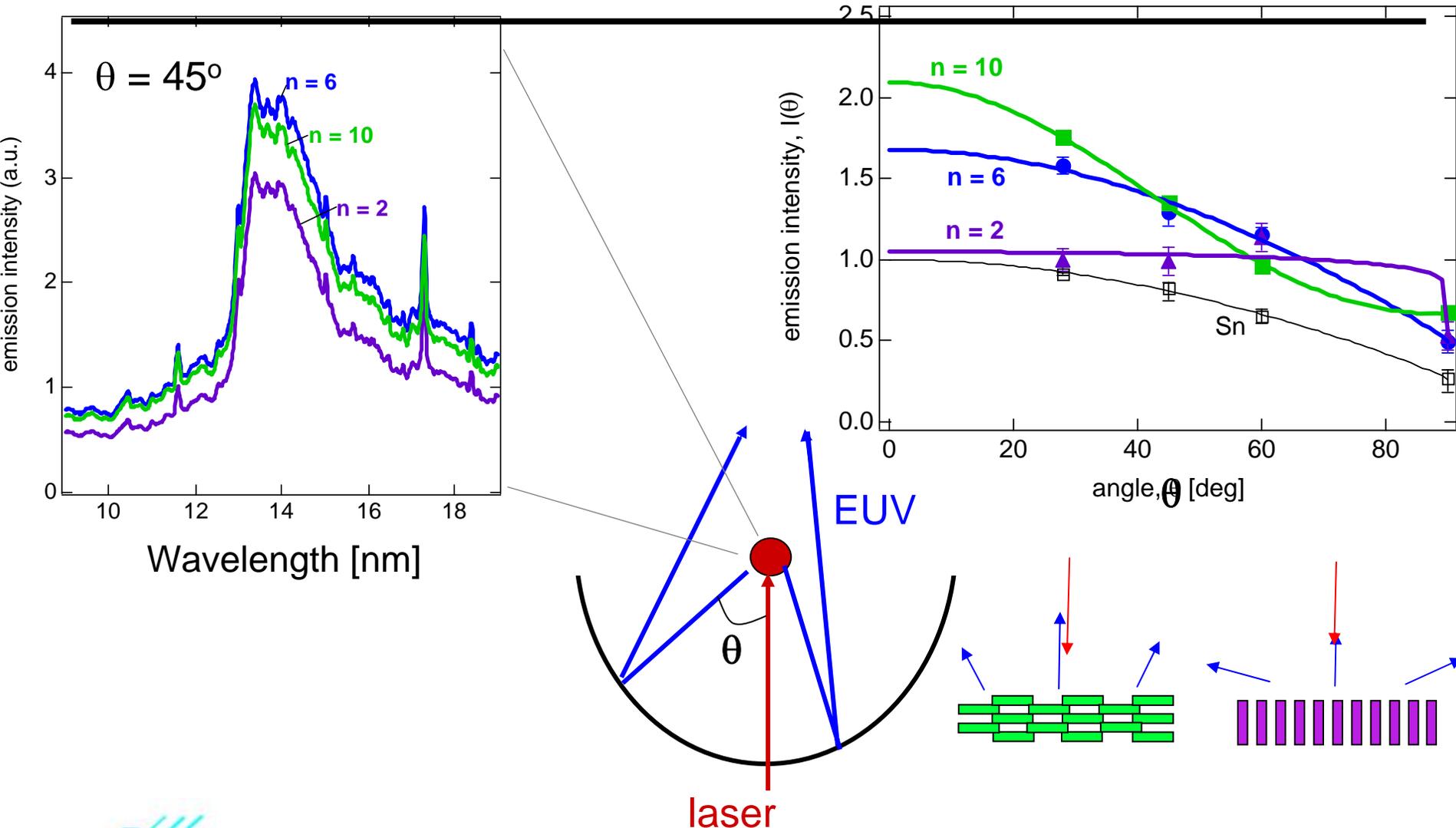
Surface template

-> Open cell

$$n = \text{EtOH}_{\text{mol}} / \text{SnCl}_{4 \text{ mol}}$$



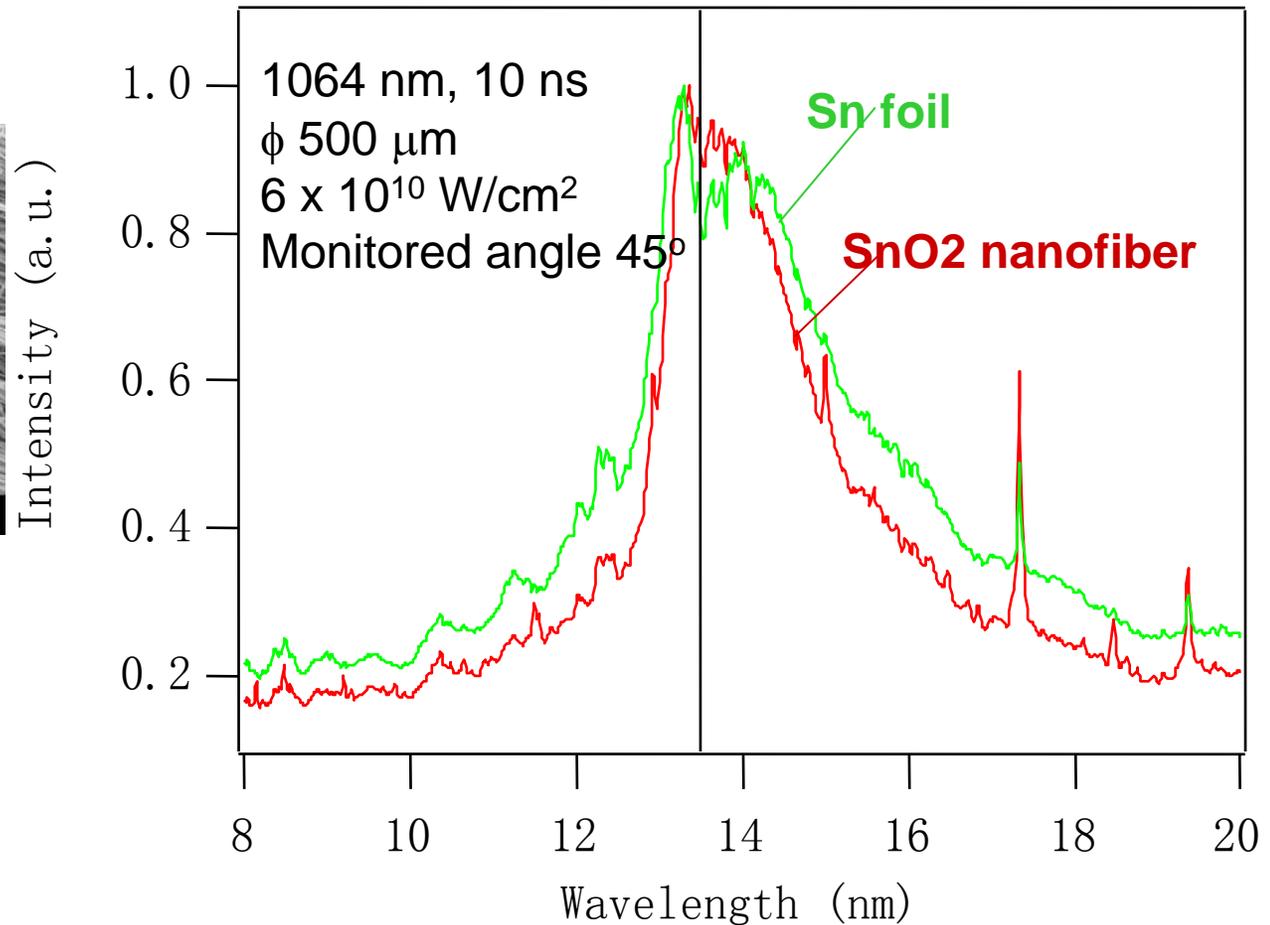
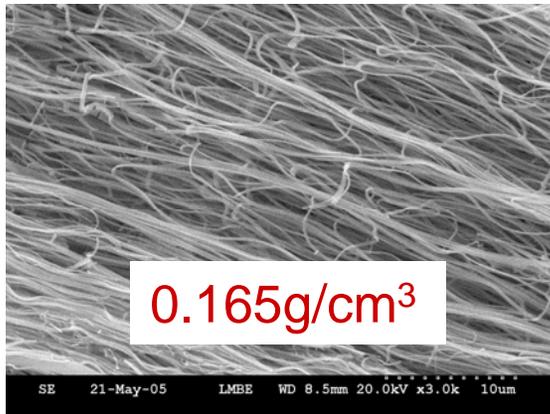
Control of angular distribution depending on target surface morphology



K. Nagai et al., *Appl. Phys. Lett.*, **88** (9), 094102, (2006).

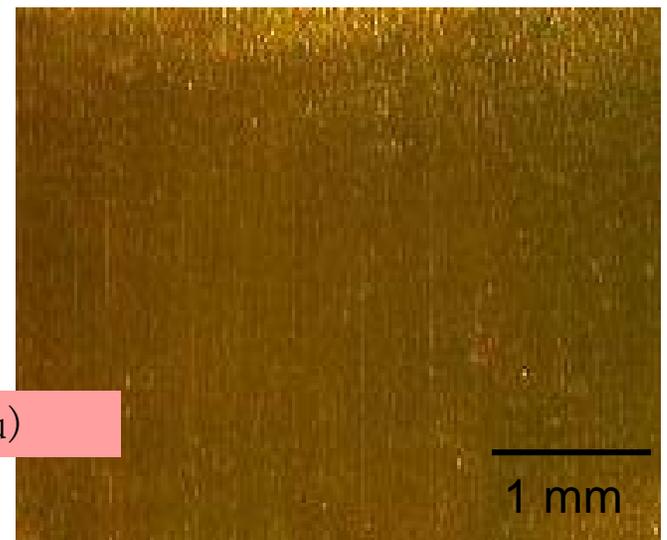
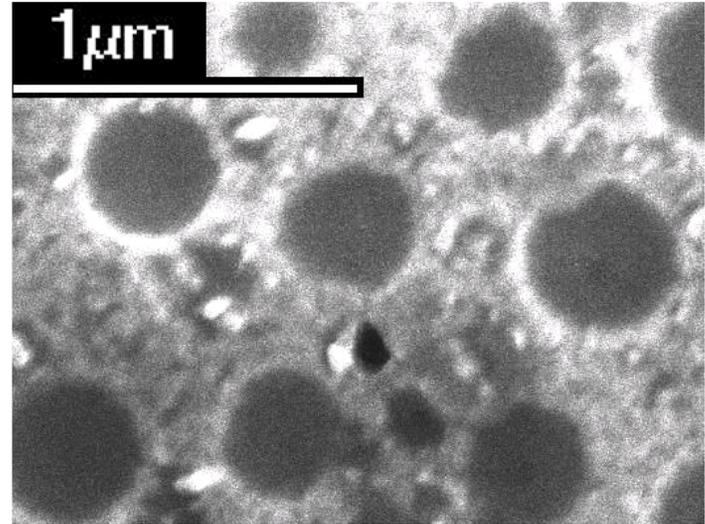
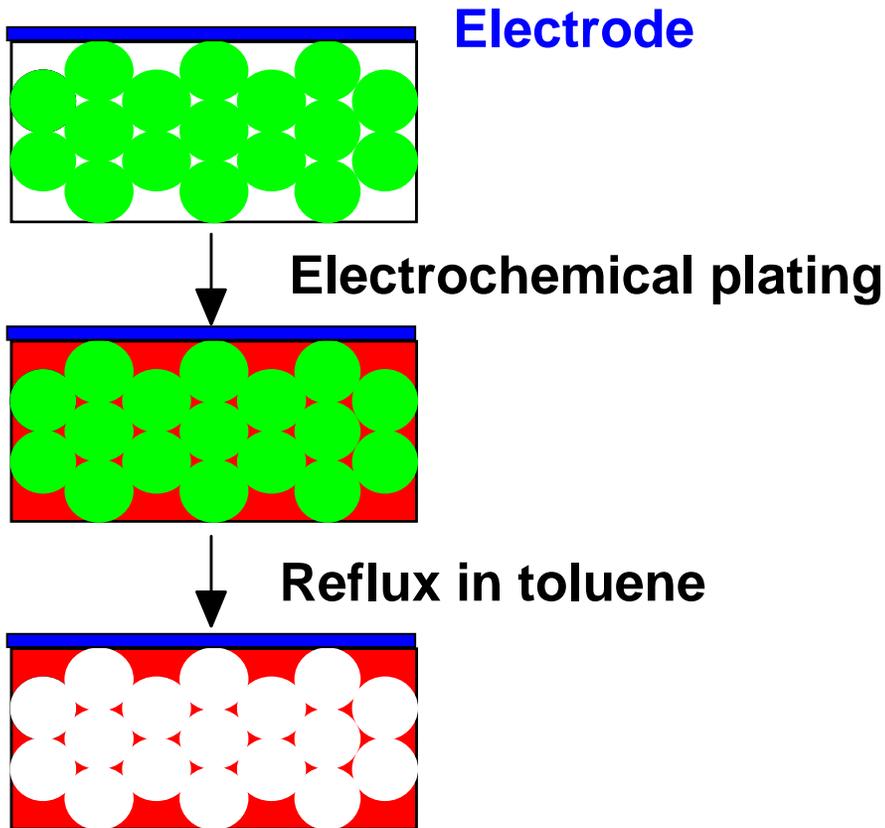


Nanofiber is an oriented low-density target.





Au foam was prepared using nanosphere template.



Thickness: 2 μm , Density: 4.3g/cm³ (23% of Au)

Fusion Sci. Technol., **49** (4), 686-690, (2006).

Specification of EUV Light source



EUV wavelength 13.5 nm with 2% band width

EUV power 115 W

Etendue 1~3.3 mm²sr

*etendue:source area x solid angle

Frequency 7~10 kHz

Stability of the power $\pm 0.3\%$ (average for 50 pulses)

Solid angle 0.03 ~ 0.2 sr

Lifetime of apparatus > 30, 000 hours (3.4 year)

Cost of Ownership

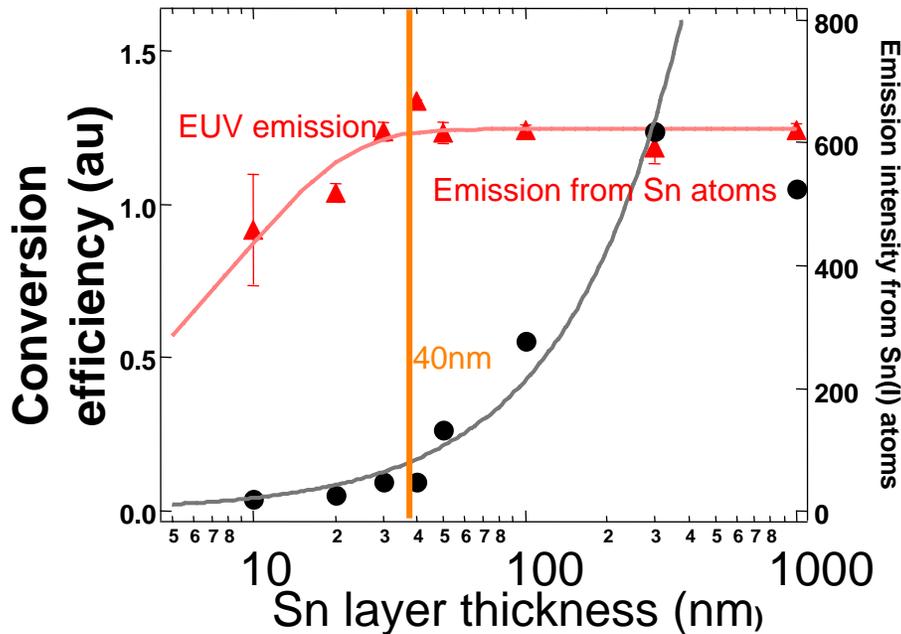


High rep irradiation requires

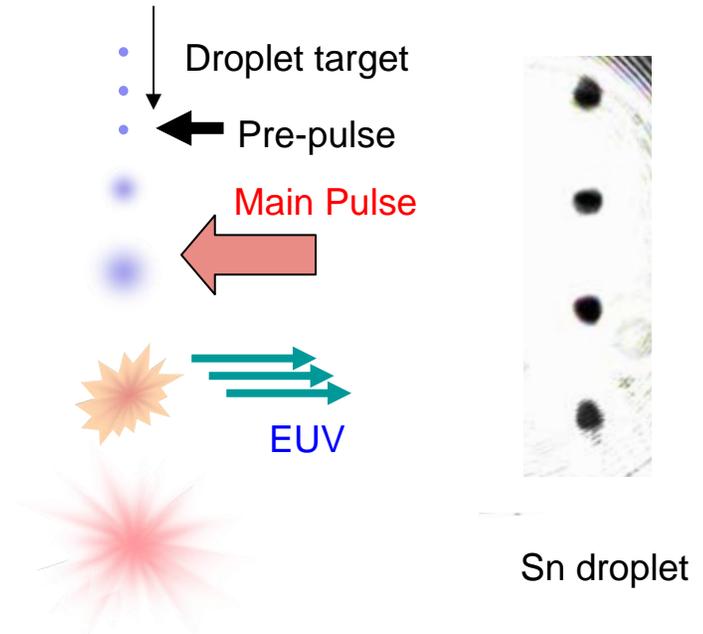


not only mass production but also minimum mass target.

Coating thickness of 40 nm is enough to produce high power EUV emission.



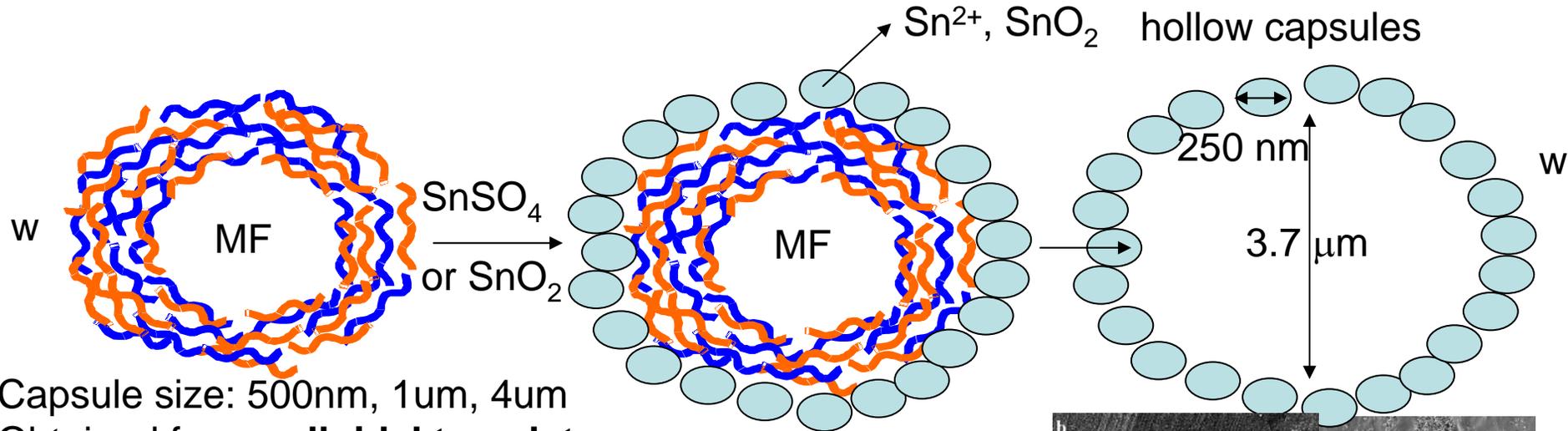
Emission from Sn neutrals linearly increases with coating thickness while keeping constant in EUV intensity.



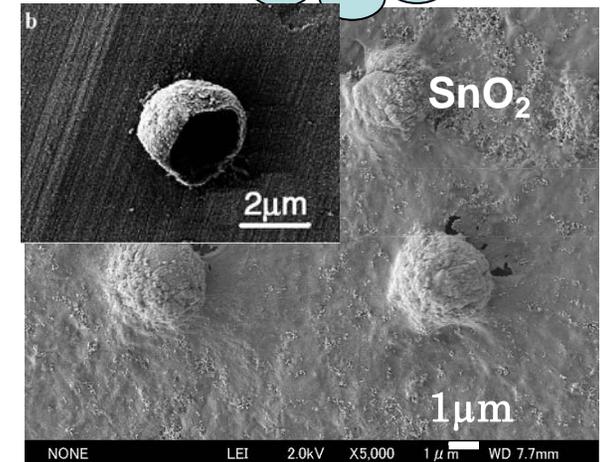
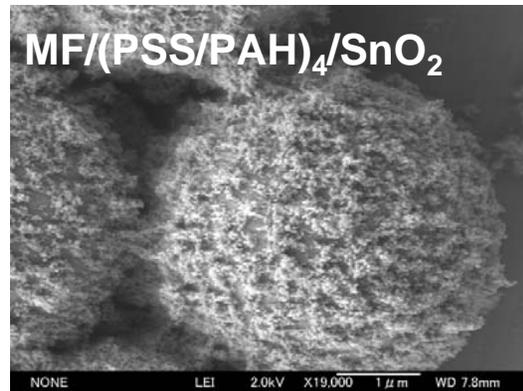
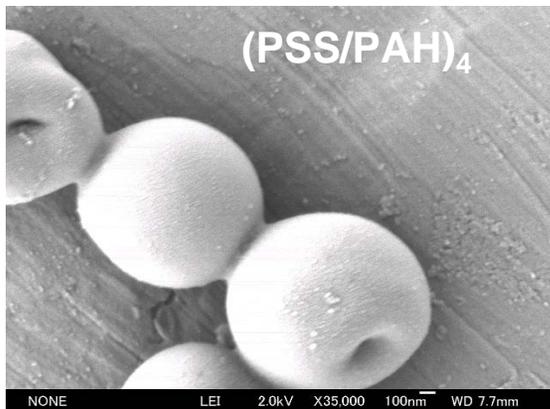
Main pulse irradiation
Intensity: $\sim 10^{11}$ W/cm²
Pulse width: ~ 10 ns

S. Fujioka et al., Appl. Phys. Lett. 87, 241503 (2005), S. Namba et al., Appl. Phys. Lett. 88, 171503 (2006).

Microbubble targets prepared from a MF template and layer by layer (LBL) coating



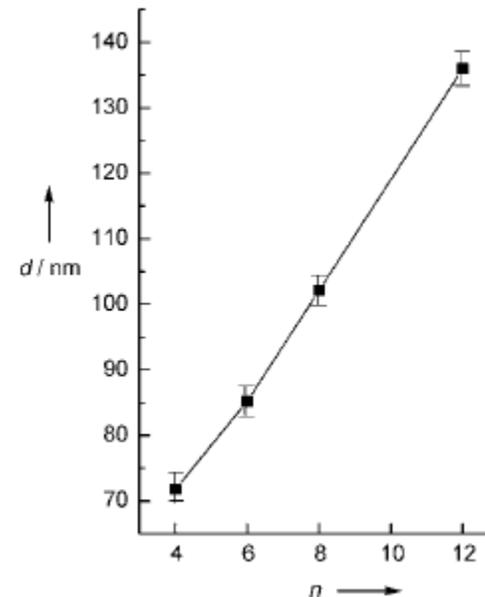
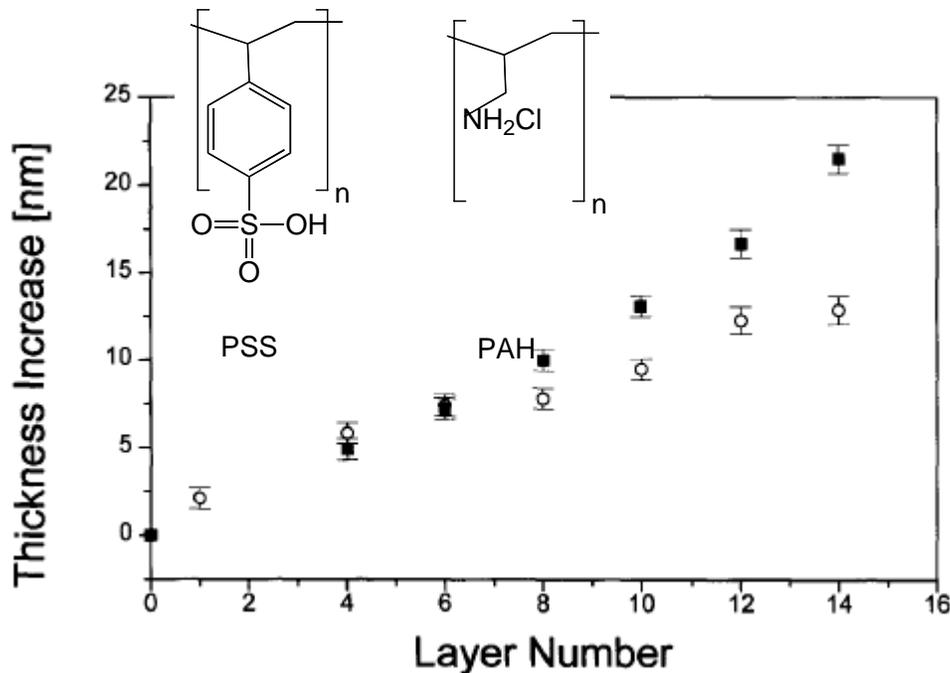
Capsule size: 500nm, 1 μm , 4 μm
Obtained from **colloidal template**



72% length (shrinkage)
14 mg/cm³ 14

LBL coating is well established technique.

The thickness of each layer has been reported to be 2~7 nm.



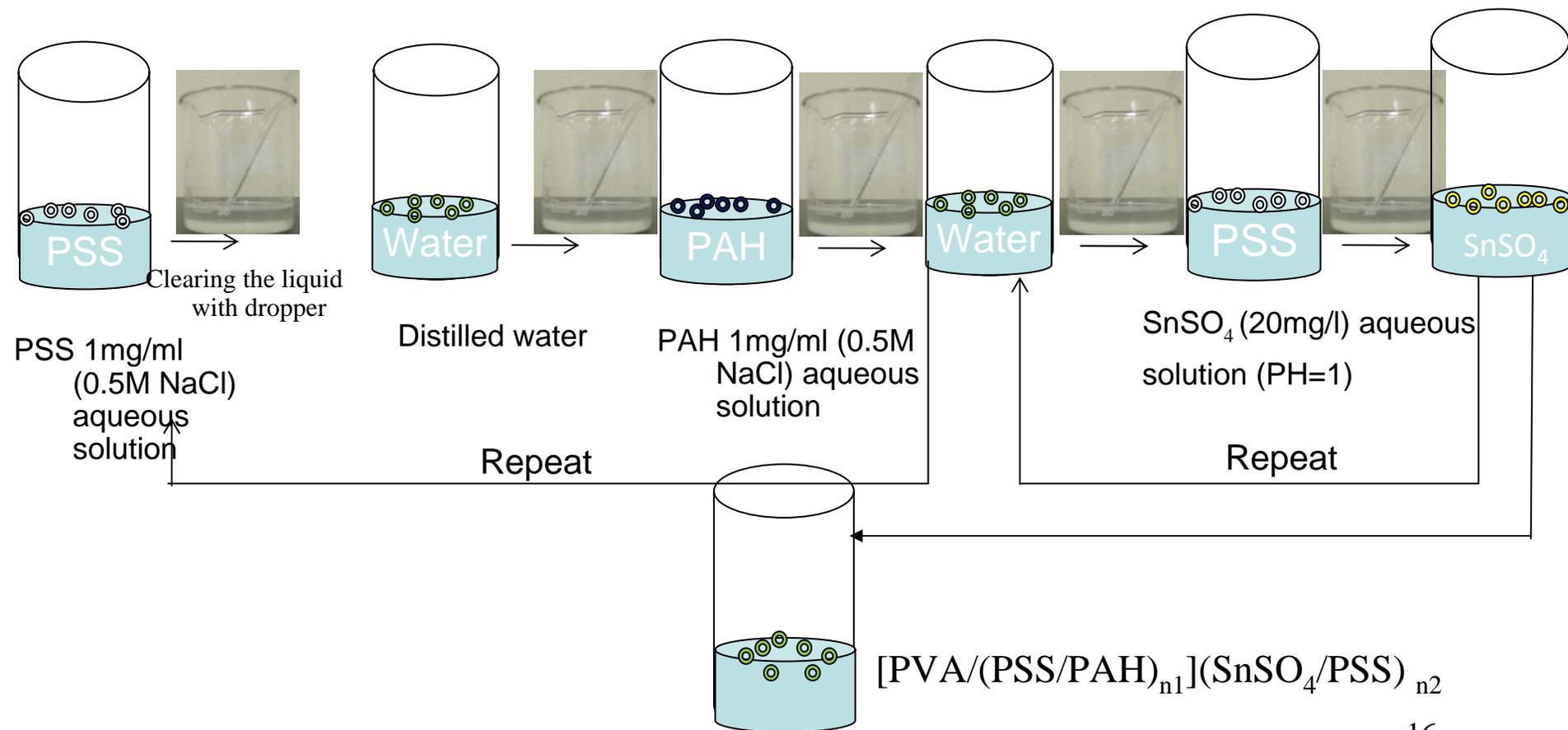
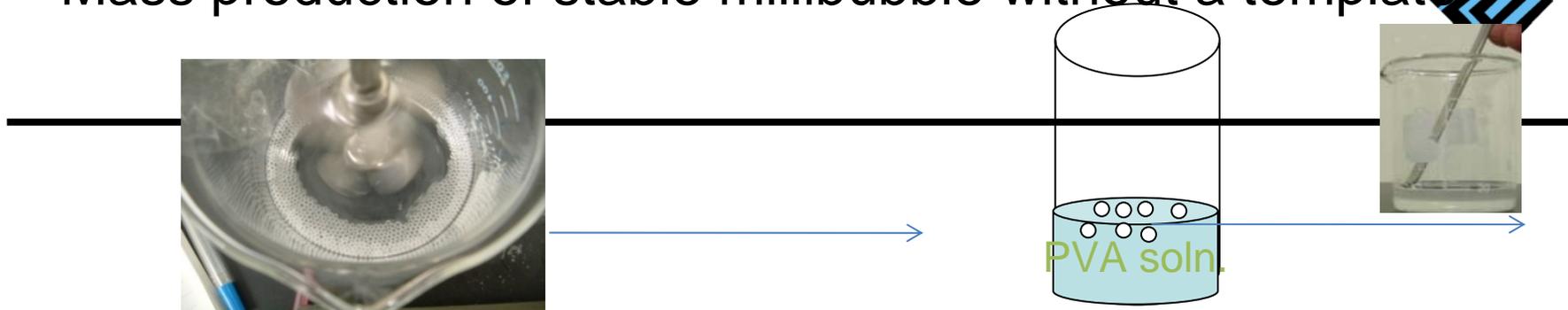
Layer thickness as measured by single particle scattered light intensity as a function of layer number. PAH/PSS coated 640 nm \varnothing polystyrene sulfate latex particles. Data calculated assuming $n_{\text{layer}} = 1.47$. • = layer deposition with the centrifugation technique; O = subsequent addition of polyelectrolyte without centrifugation.

The relationship between the thickness d and the layer number n .
(tween:span/((PAH/PSS) $_n$)

Angew. Chem. Int. Ed. 2005, 44, 3310–3314

Colloids Surfaces A: Physicochem. Eng. Aspects 137 (1998) 253-266

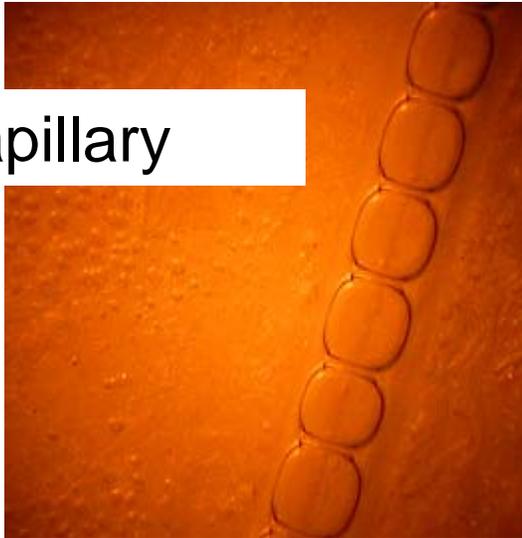
Mass production of stable millibubble without a template



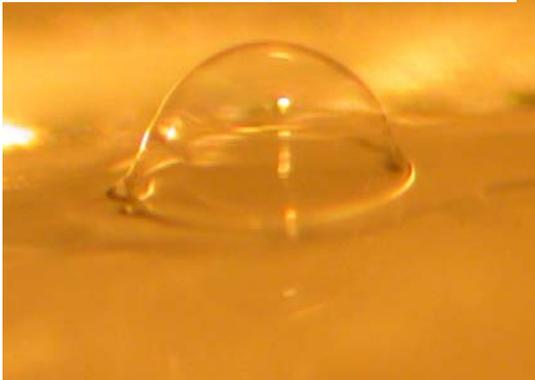
Tin coated millibubble



•in a capillary



•Dried on a glass slide



The amount of tin atom per one capsule was estimated to be

0.9×10^{15} atom for
[PVA/(PSS/PAH)₃](Sn²⁺/PSS)₃

3.2×10^{15} atom for
[PVA/(PSS/PAH)₃](SnO₂/PSS)₃

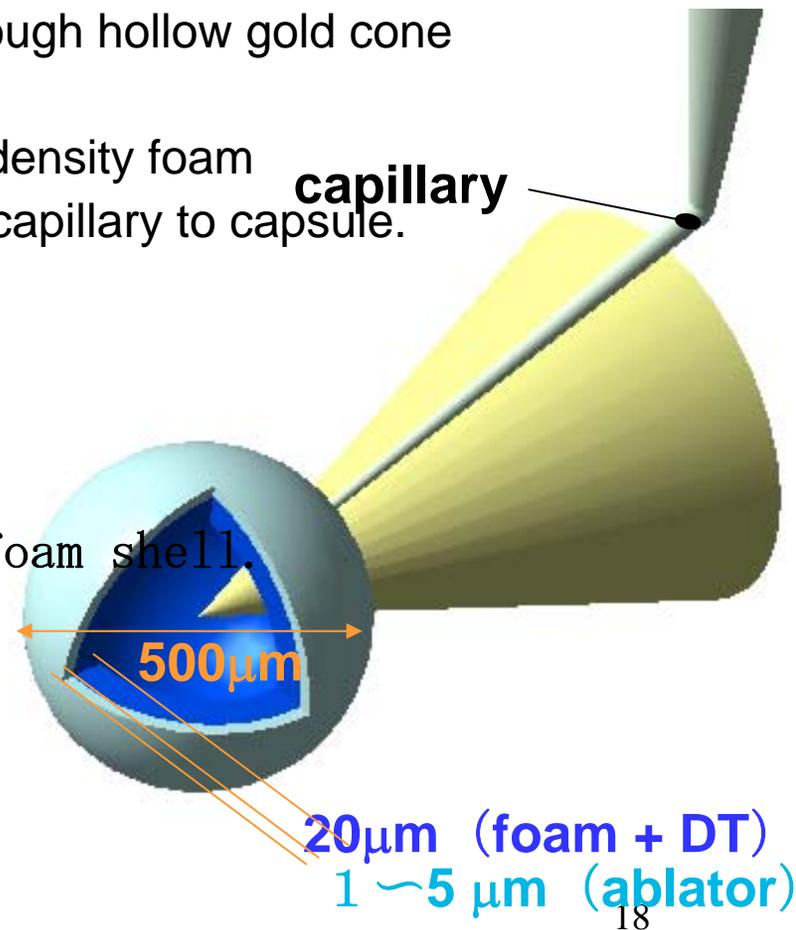
Low density materials for cryogenic foam targets



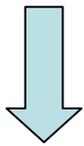
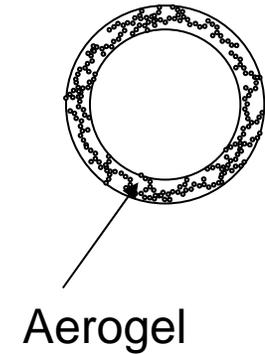
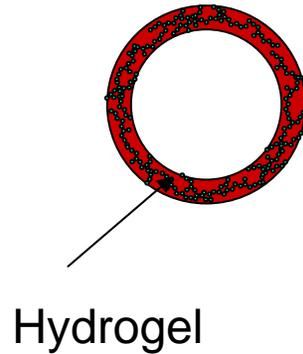
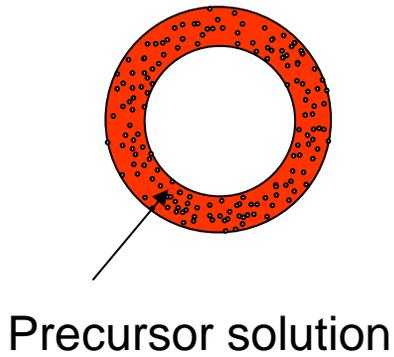
- Compression of fuel (spherical DT) using Gekko XII laser.
- Heating by LFEX (new petawatt) laser through hollow gold cone
- The fuel capsule was prepared using low density foam and liquid DT will be infiltrated through an capillary to capsule.

Specification

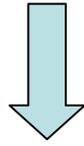
- D_2 or DT fuel is infiltrated into a foam shell.
- Shell diameter : $500 \mu\text{m}$
- Thickness of DT : $\sim 20 \mu\text{m}$
- Ablator thickness : $\sim 5 \mu\text{m}$



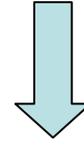
Aerogel capsule fabrication -4 steps-



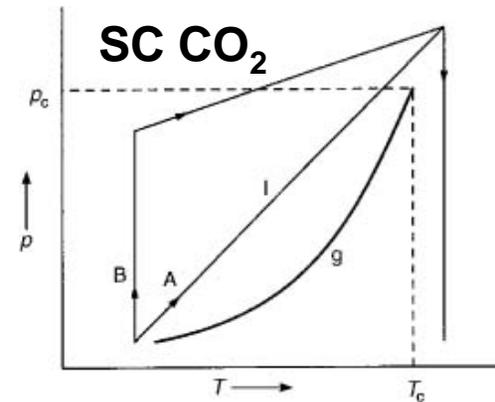
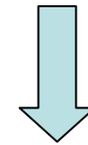
Droplet generator



Phase transfer catalyst



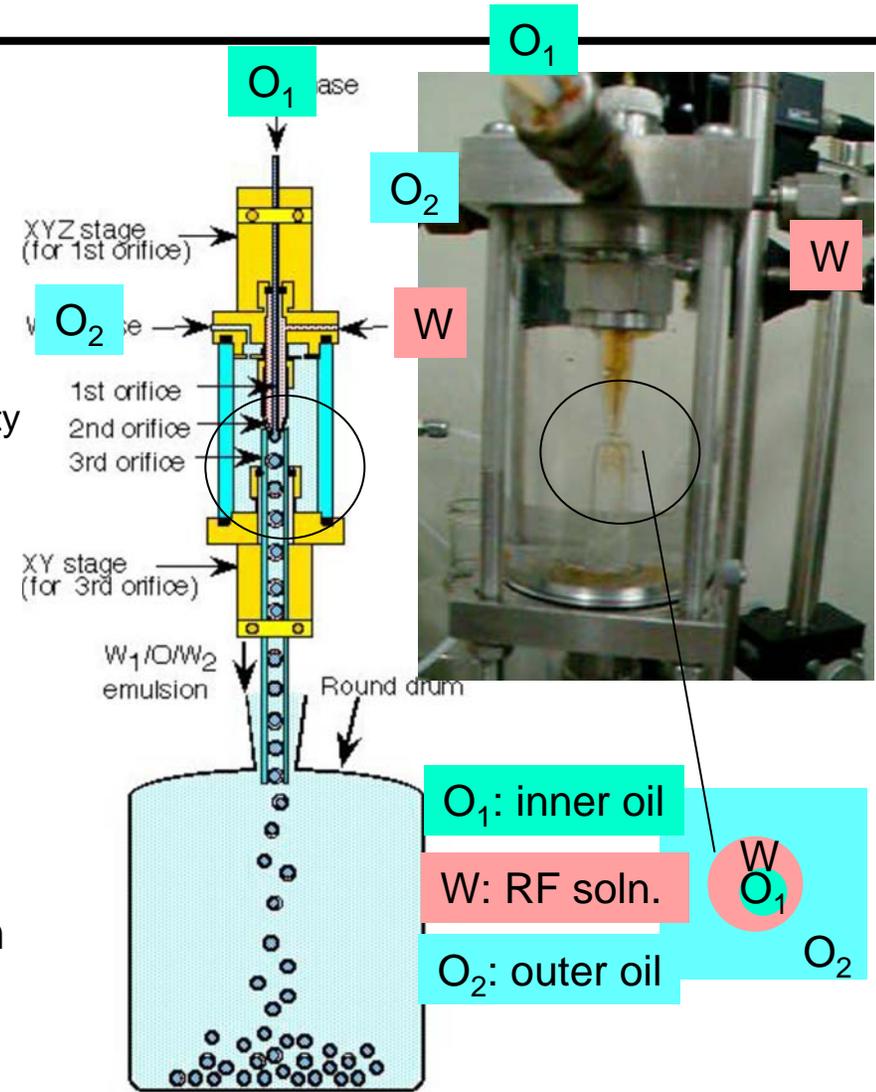
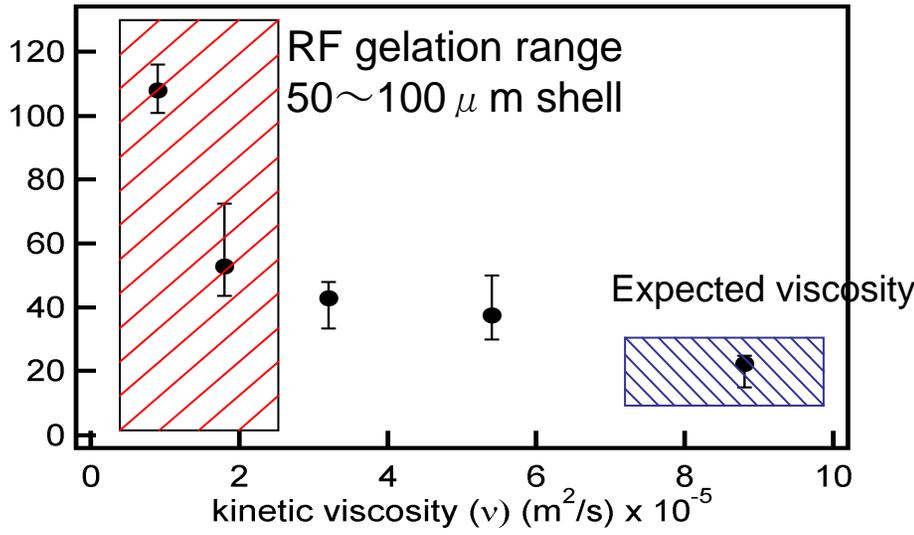
Porous structure control



 aqua  aqua  organic solvent  gas



The thickness of W depends on its viscosity.

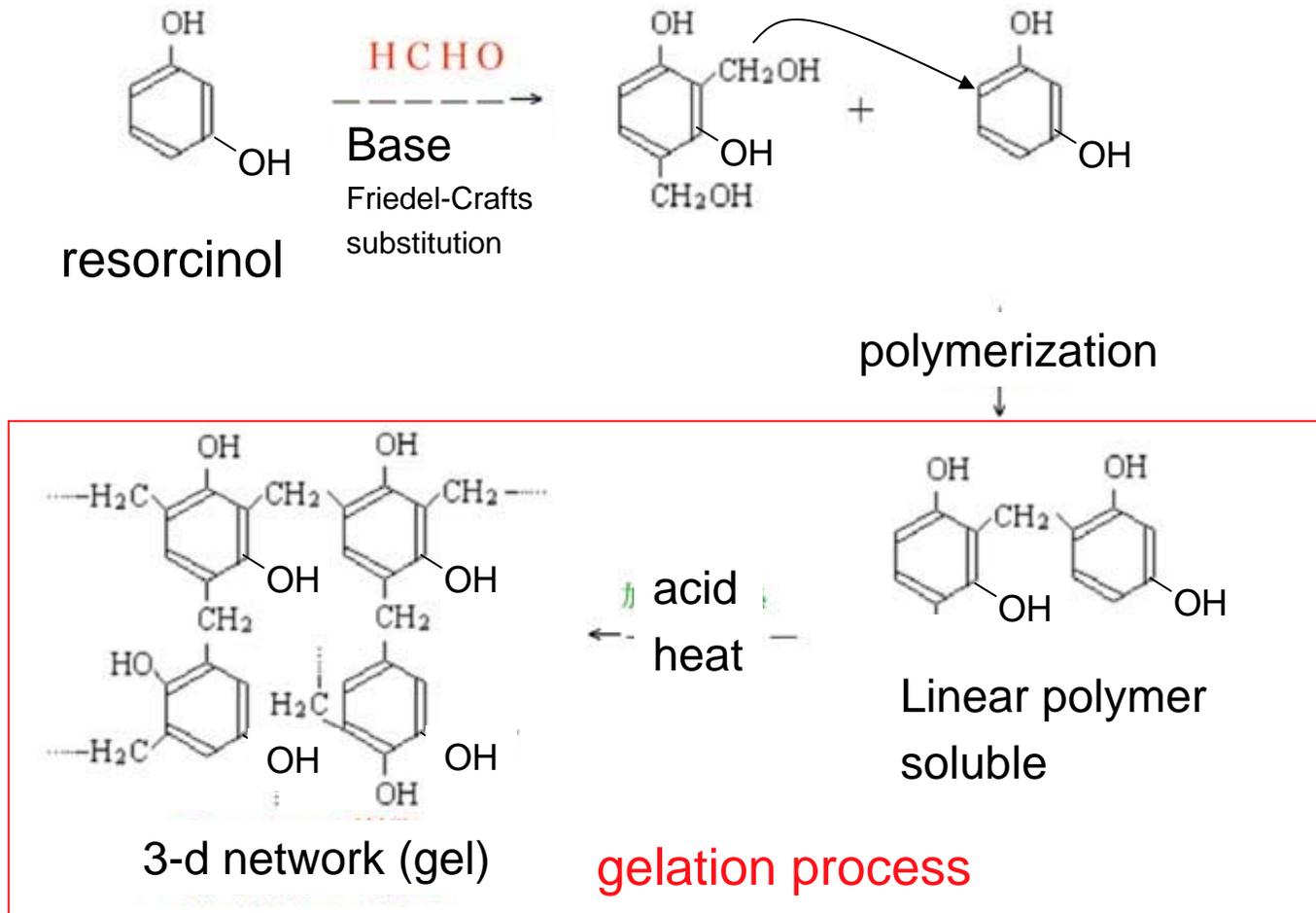


- The high viscosity induced thin wall thickness.
- Viscous RF solution is high concentration and cannot be applied for the emulsion process due to gelation.

Gelation process is key issue to control density and capsule morphology, and it is chemical process of crosslinking of polymer chain.

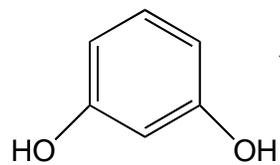


• In the case of RF,

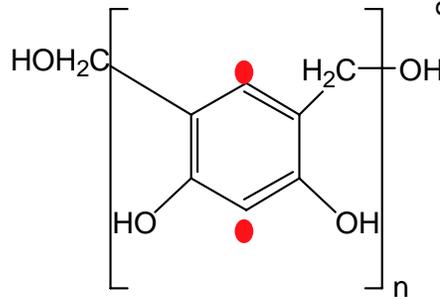




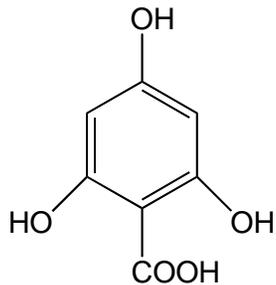
Linear polymer without crosslinker (•) increase its viscosity.



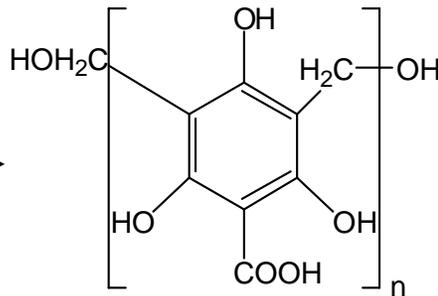
resorcinol



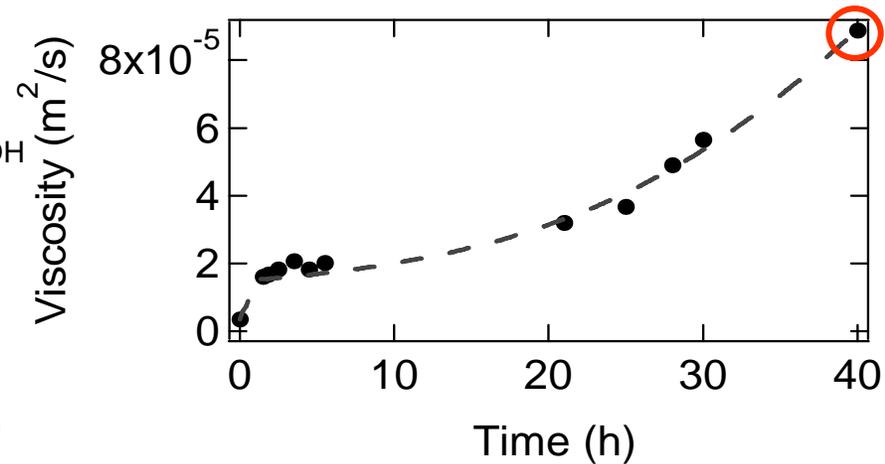
resorcinol/formalin (RF)
(linear)



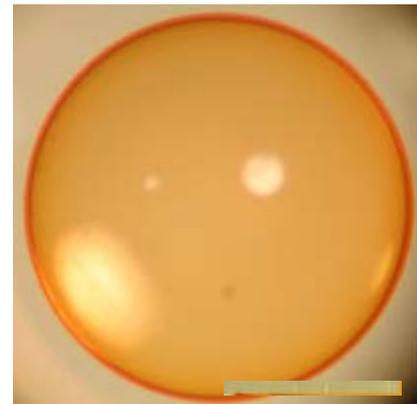
phloroglucinolcarboxylic acid



phloroglucinolcarboxylic acid)/formalin (PF)



Polymerization of PF solution

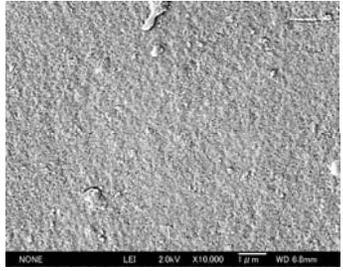
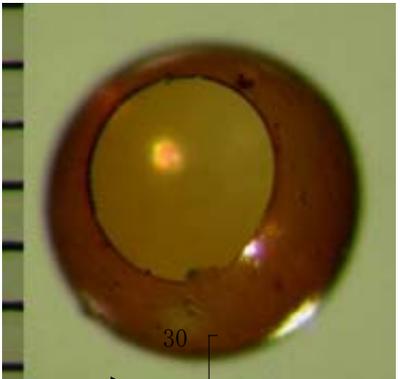
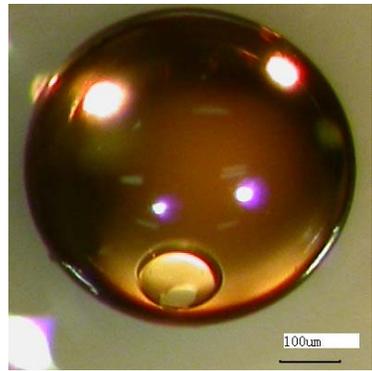


Target assembly



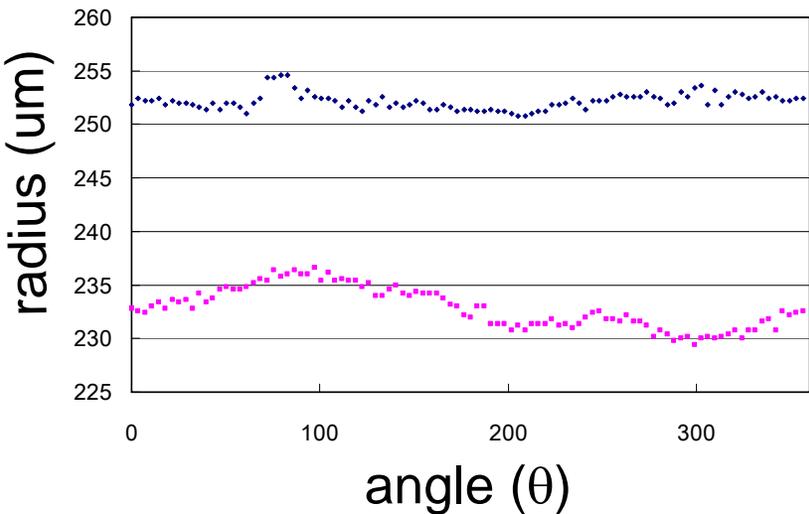
RF/PF foam shell

radius $252 \pm 0.3 \mu\text{m}$
 Wall thickness $19.3 \pm 1.3 \mu\text{m}$

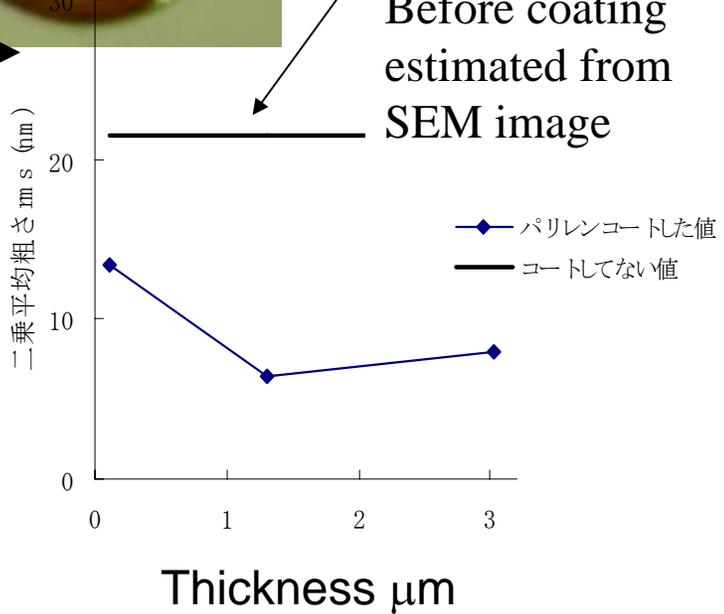


Hole drilling

Before coating estimated from SEM image



Inner and outer radius of foam layer depending on the angle.



Roughness estimated from AFM images depending on the ablator thickness₂₃

Cryogenic fueling test

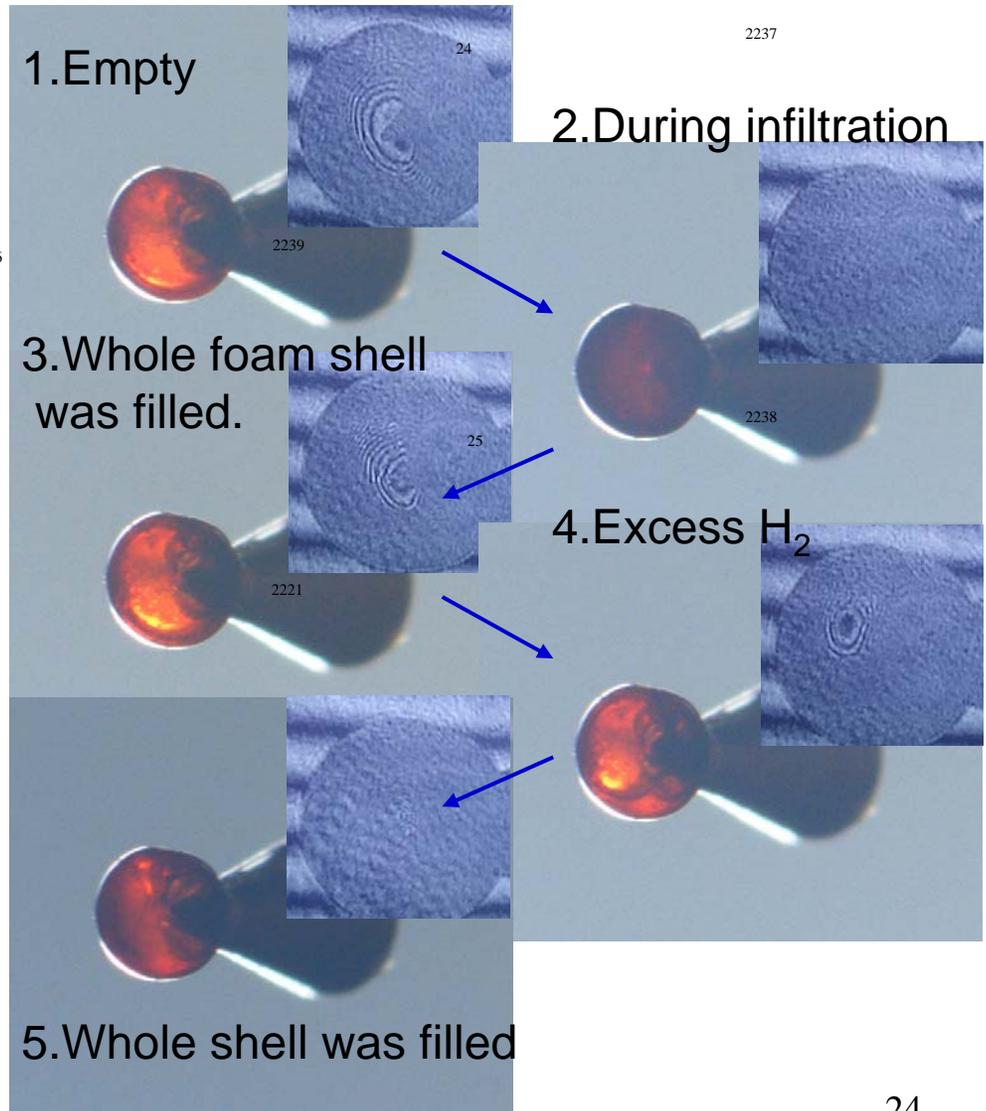


20

2234

conditions

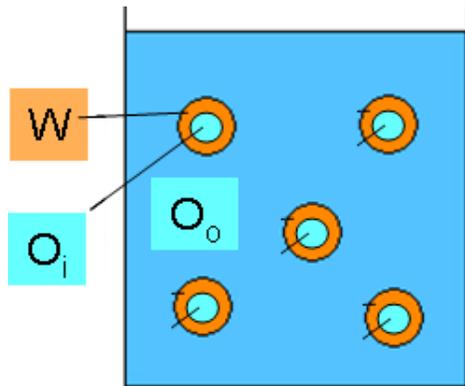
- gas temperature : 12.5 K
- H₂ pressure : 7.3 kPa



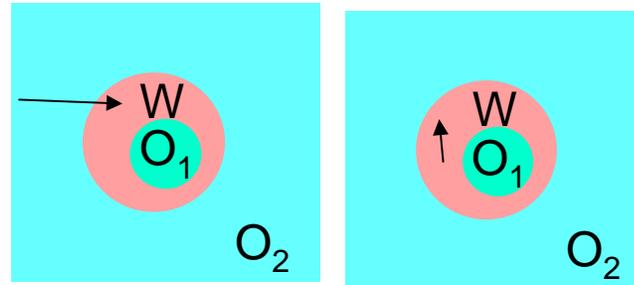
Iwamoto, Fusion Eng. Design
2007



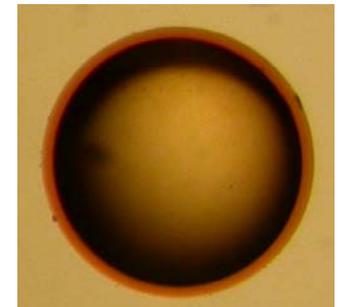
Phase transfer catalyst to keep density matching.



- O_o: silicon oil mixture**
 $\rho = 1.018 \text{ g/cm}^3$
- O_i: 1-methylnaphathlene**
 $\rho = 1.02 \text{ g/cm}^3$
- W: PF aqueous solution**
 $\rho = 1.02 \text{ g/cm}^3$

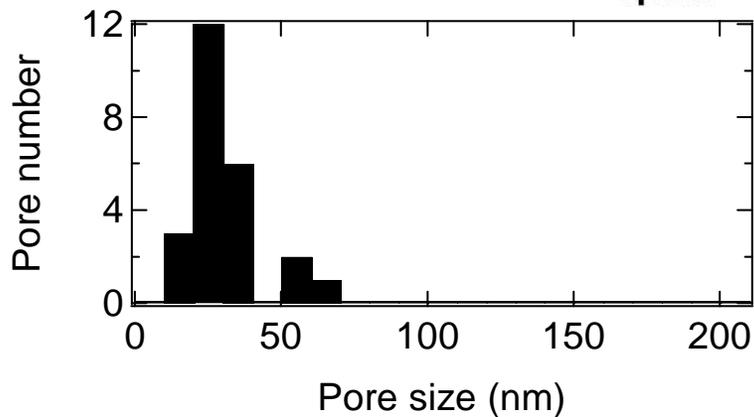
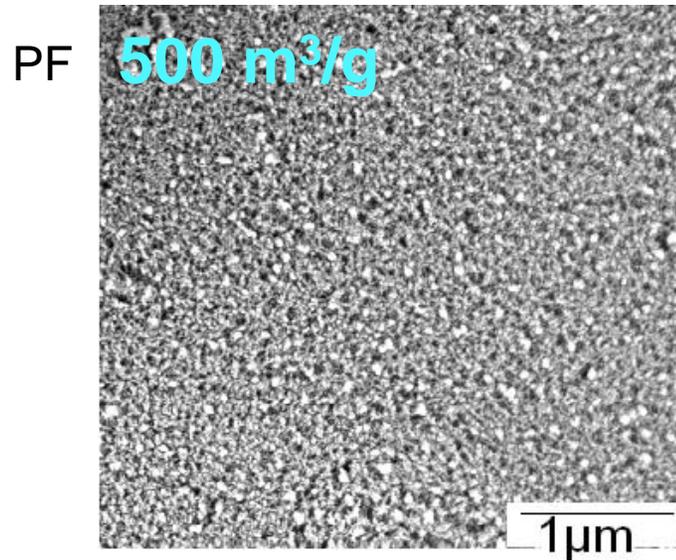


A phase transfer catalyst for gelation transfers from O to W (RF solution), while the previous catalyst for gelation was dissolved in W (RF solution) phase, and activated by heating.

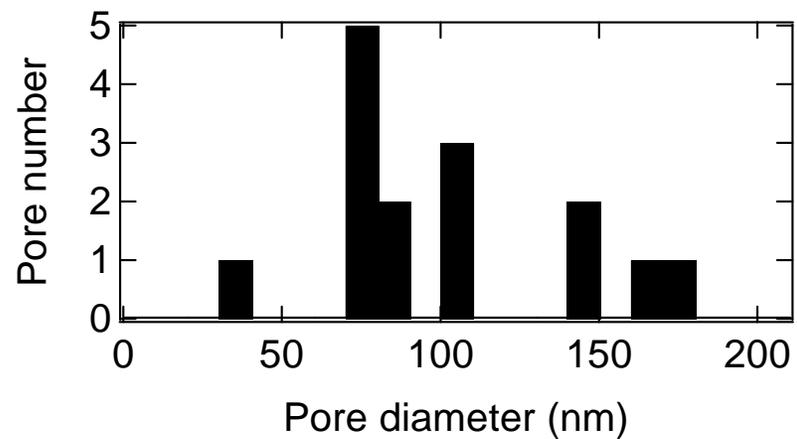
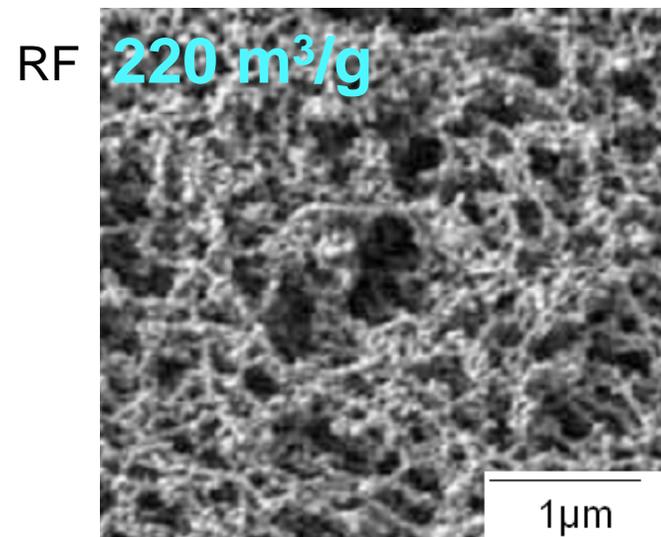


- F. Ito, K. Nagai, M. Nakai, and T. Norimatsu: *Macromol. Chem. Phys.* **206** (2005) 2171.
F. Ito, K. Nagai, M. Nakai, and T. Norimatsu, *Fusion Sci. Technol.* **49** (2006) 663-668.

PF aerogel has finer pore than RF aerogel (32 ± 10 nm) with smaller distribution.



Particle network

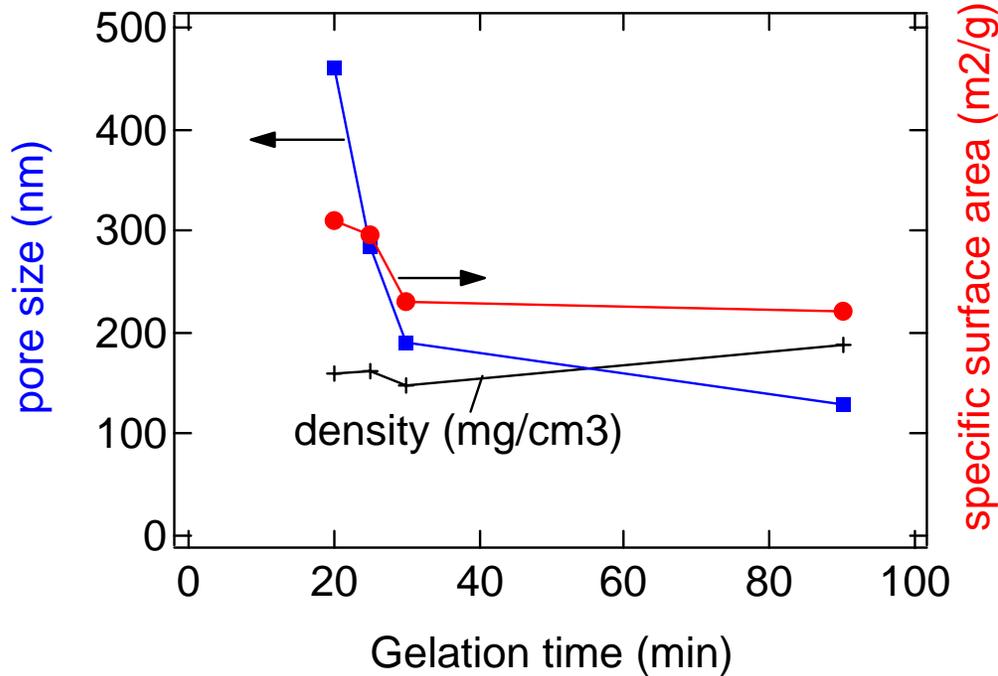


Fiber network

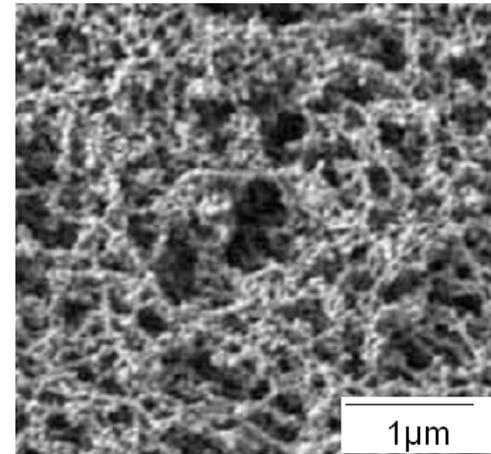


Network size:

little nuclear, fast particle growth -> high pore size, high surface area



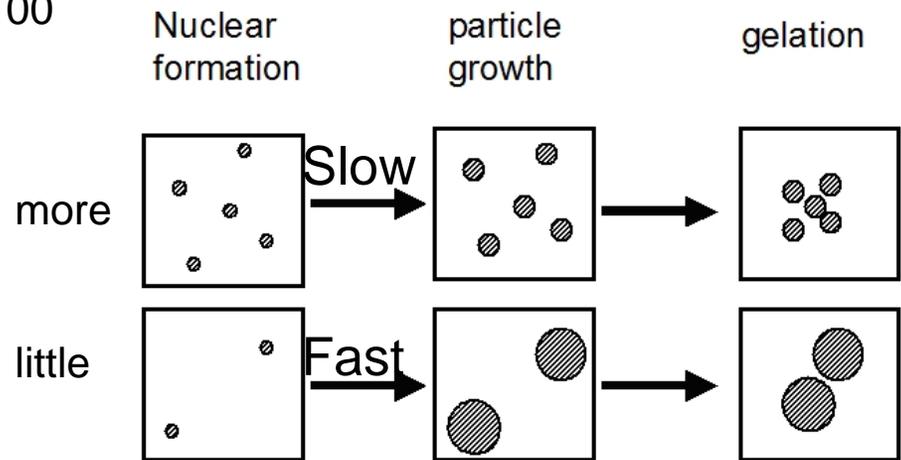
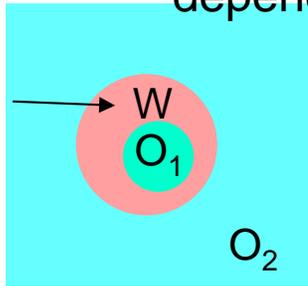
RF



Characterization of RF aerogel depending on gelation time.

The gelation time was controlled by catalysts.

F. Ito, et al., *Macromol. Chem. Phys.*, **206** (21), 2171-2176, (2005).



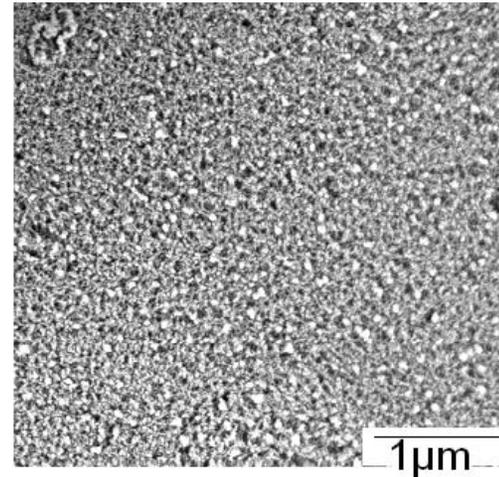
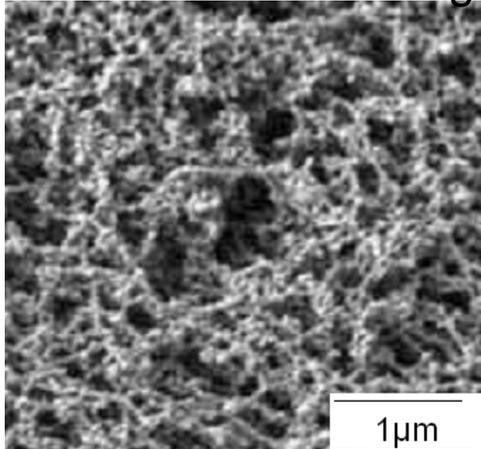


Network structure:

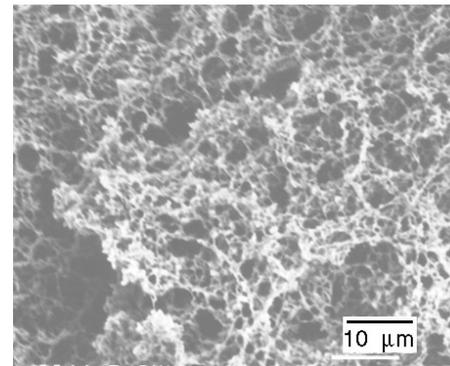
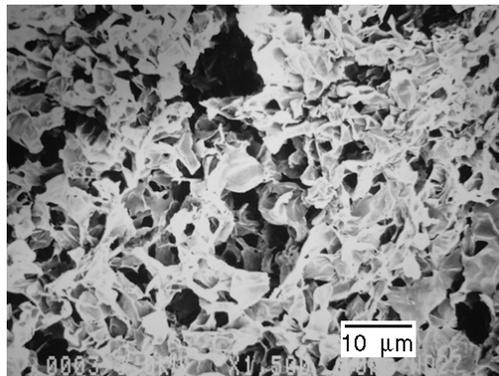
Labile bond -> particle like (reorientation of linkage)

RF: covalent bond linkage (inert)

PF: hydrogen bond linkage (labile)



TMP (TPX) : van der Waars interaction (labile)



Gelation in a hexanol

Gelation in a butanol

Summary



1. Low density Sn and Au using template, electrospinning
2. High rep target ----necessity of minimum mass target
3. Organic aerogel capsule
4. Discussion of nanostructure control